

SHRIMATHI DEVKUNVAR NANALAL BHATT VAISHNAV COLLEGE FOR WOMEN
(AUTONOMOUS)

(Affiliated to the University of Madras and Re-accredited with 'A+' Grade by NAAC)
Chromepet, Chennai - 600 044.

M.Sc.Chemistry - END SEMESTER EXAMINATIONS - APRIL 2025

SEMESTER - II

22PCHCT2004 - Organic Reaction Intermediates and its Mechanism

Total Duration : 2 Hrs. 30 Mins.

Total Marks : 60

Section B

Answer any **SIX** questions ($6 \times 5 = 30$ Marks)

1. Explain any two factors responsible for the stability of carbocation.
2. Illustrate the role of chloroform and alkali in Reimer-Tiemann reaction.
3. Predict the products and provide the mechanism with stereo chemistry.

$$\text{CH}_3\text{CHBrCHBrCH}_3 \xrightarrow[\text{Acetone}]{\text{Zn}} ?$$
4. Discuss Sharpless asymmetric epoxidation with suitable examples.
5. Predict the product, give the mechanism and account for the stereo chemical course of the reaction. $\text{PhCHBrCH}_3 \xrightarrow[\text{CH}_3\text{COOH}]{\text{CH}_3\text{COOAg}} ?$
6. Apply the orientation effect and explain why the nitration of toluene gives o- and p-nitrotoluene readily and predominantly?
7. Discuss the mechanism of Michael reaction.
8. Examine the hydroxylation reactions of alkene with OsO_4 and KMnO_4 .

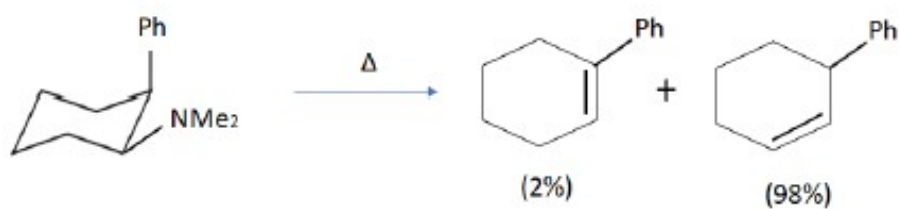
Section C

I - Answer any **TWO** questions ($2 \times 10 = 20$ Marks)

9. What are free radicals? How are they classified? Give any two methods for the generation of free radicals.
10. Explain the following conversions with mechanism.
 - (i) 2 - aminopyridine from pyridine.
 - (ii) 2 - carbethoxy cyclopentanone from ethyl adipate.
11. (i) Justify the fact that the electrophilic aromatic substitution reactions occur by a stepwise mechanism.
 (ii) Differentiate aromatic electrophilic substitution from aromatic nucleophilic substitution.

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12. (i) Distinguish the factors controlling the competing reactions of substitution and elimination in alkyl halides.
(ii) Justify the following observation.



II - Compulsory question (1 × 10 = 10 Marks)

13. Discuss the synthetic applications of the following.
(i) LTA (ii) SeO₂ (iii) Birch reduction.
